



Belfairs Academy

A Level Chemistry Fundamentals Map

| Knowledge | Skills |
|--|---|
| DEVELOPMENT OF PRACTICAL SKILLS IN CHEMISTRY Experimental design, including to solve problems set in a practical context Processing, analysing and interpreting qualitative and quantitative experimental results Use and application of scientific methods and practices Research and referencing | Work to significant figures Draw and interpret graphs Use ratios Use estimates Use standard form Determine the slope and intercept of a linear graph Translate information between graphical, numerical and algebraic forms |
| FOUNDATIONS IN CHEMISTRY The mole and calculations using moles Oxidation numbers and redox reactions Acid-base reactions Electronic configuration and chemical properties | Recognise and make use of appropriate units in calculations Use expressions in decimals and ordinary form Use ratios, fractions and percentages |
| PERIODIC TABLE AND ENERGY Properties and reactions of Group 2 elements and the halogens The periodic table and periodicity Patterns in the periodic table Bonding Intermolecular forces Shapes of molecules and ions Energetics and enthalpy changes Rates of reactions Chemical Equilibria | Solve algebraic equations Change the subject of an equation |
| CORE ORGANIC CHEMISTRY Introductory Organic chemistry Free radical substitution reactions of alkanes Electrophilic addition reactions of alkenes Isomerism in alkanes and alkenes Addition polymerisation Alcohols and their reactions Haloalkanes and their reactions Mass spectra Infrared spectroscopy Organic synthesis | Use expressions in decimals and ordinary form Use ratios, fractions and percentages |



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PHYSICAL CHEMISTRY AND TRANSITION ELEMENTS

Measuring reaction rates
Orders and the rate equation
Half-lives
Orders from rate-concentration graphs
Initial rates and the rate constant
Rate-determining step
The equilibrium constants- K_c and K_p
Calculations using K_c and K_p
What is pH?
Conjugate acid-base pairs
Calculating pH for strong and weak acids
The ionisation of water and pH of bases
Buffer solutions and calculating their pH
Titration curves
Lattice enthalpy
Constructing Born-Haber cycles and related calculations
Enthalpy change of solution and hydration
Entropy and free energy
Redox
Half cells and cell potentials
The feasibility of reactions
Transition metals and complex ions
Stereoisomerism in complex ions
Ligand substitution in complexes
Redox titrations

Recognise and make use of appropriate units in calculations
Estimate results
Use calculators to find and use power, exponential and logarithmic functions
Translate information between graphical, numerical and algebraic forms



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| <p>ORGANIC CHEMISTRY AND ANALYSIS</p> <p>Introduction to aromatic chemistry</p> <p>The structure of benzene</p> <p>The delocalised model of benzene</p> <p>Substitution reactions of benzene</p> <p>Phenols</p> <p>An introduction to carbonyl compounds</p> <p>Reactions of aldehydes and ketones</p> <p>Carboxylic acids</p> <p>Esters</p> <p>Amino acids and amines</p> <p>Addition and condensation polymerisation</p> <p>Organic synthesis of aliphatic and aromatic compounds</p> <p>Thin-layer chromatography</p> <p>Gas Chromatography- mass spectrometry</p> <p>Carbon-13 nuclear magnetic resonance</p> <p>Proton nuclear magnetic resonance</p> | <p>Use expressions in decimals and ordinary form</p> <p>Use ratios, fractions and percentages</p> |